



Ozone production

Adapted from NCRP Report 51 and LEP Note 379 (via APresland 2005-10-10.pdf), under the assumption of no O_3 decomposition (yielding in the τ expression a neglected term $kP_{\rm eV}/V$ with k decomposition constant equal to 1.4 10^{-16} cm³/eV):

$$C_{O_3} = \frac{C_{O_2}GP_{eV}\tau}{N_{Av}\left(\frac{\rho_{Air}V}{A_{Air}}\right)} \left(1 - e^{-\frac{t}{\tau}}\right) \qquad \tau = (\alpha + 1/\tau_{vent})^{-1} \quad \alpha = 2.3 \cdot 10^{-4} \,[\text{s}^{-1}] \quad O_3 \text{ dissociation constant}$$

$$C_{O_2} = 0.232$$
 $G = 0.06 - 0.074$ $[O_3 / eV]$ $N_{Av} \frac{\rho_{Air}}{A_{Air}} @ NTP = 2.50 \cdot 10^{19} [molecules/cm^3]$

$$P_{eV}$$
 [eV/s] = 6.24·10¹⁸ P [W] $r = \frac{1}{\tau_{vent}}$ [air renewal/s]

$$P_{eV} [\text{eV/s}] = 6.24 \cdot 10^{18} P[\text{W}] \qquad r = \frac{1}{\tau_{vent}} [\text{air renewal/s}]$$

$$C_{O_3} [\text{ppm}] = 9.28 \cdot 10^{-15} G[\text{eV}^{-1}] \frac{P_{eV} [\text{eV/s}] \tau[s]}{V[\text{cm}^3]} \left(1 - \text{e}^{-\frac{t}{\tau}}\right)$$